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## Structure Reports

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**(*E,E*)-4,4'-Dichloro-2,2'-(1,1'-azino-diethylene)diphenol**Jian-Guo Chang,<sup>a\*</sup> Dong-Feng Zhao,<sup>b</sup> Guo-Fang He<sup>a</sup> and Ji-Kun Li<sup>a</sup><sup>a</sup>Department of Materials Science and Chemical Engineering, Taishan University, 271021 Taian, Shandong, People's Republic of China, and <sup>b</sup>Department of Chemical Engineering, Weifang Vocational College, 261041 Weifang, Shandong, People's Republic of China

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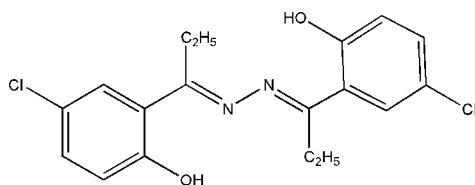
Received 14 June 2007; accepted 20 June 2007

Key indicators: single-crystal X-ray study;  $T = 273$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.032;  $wR$  factor = 0.089; data-to-parameter ratio = 13.6.

The title compound,  $\text{C}_{18}\text{H}_{18}\text{Cl}_2\text{N}_2\text{O}_2$ , was synthesized by the reaction of 1-(5-chloro-2-hydroxyphenyl)propan-1-one with hydrazine hydrate. The molecule possesses a crystallographically imposed centre of symmetry at the midpoint of the N—N bond. Intramolecular O—H...N hydrogen bonds, together with C—H... $\pi$  interactions, help to stabilize the structure.

## Related literature

For related literature, see: Glaser *et al.* (1995); Hunig *et al.* (2000); Kesslen *et al.* (1999); Kundu *et al.* (2005); Zheng *et al.* (2005).



## Experimental

## Crystal data

 $\text{C}_{18}\text{H}_{18}\text{Cl}_2\text{N}_2\text{O}_2$  $M_r = 365.24$ Monoclinic,  $P2_1/c$  $a = 9.3706$  (3) Å $b = 13.9089$  (4) Å $c = 6.7640$  (2) Å $\beta = 103.021$  (1)° $V = 858.92$  (4) Å<sup>3</sup> $Z = 2$ Mo  $K\alpha$  radiation $\mu = 0.39$  mm<sup>-1</sup> $T = 273$  (2) K

0.33 × 0.23 × 0.11 mm

## Data collection

Bruker APEXII CCD area-detector diffractometer

Absorption correction: multi-scan (SADABS; Sheldrick, 2003)

 $T_{\min} = 0.882$ ,  $T_{\max} = 0.958$ 

9653 measured reflections

1507 independent reflections

1352 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.022$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.032$  $wR(F^2) = 0.089$  $S = 1.08$ 

1507 reflections

111 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.17$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.29$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1-H1\cdots N1$	0.82	1.83	2.5513 (16)	146
$C2-H2\cdots Cg1^i$	0.93	2.68	3.496 (2)	147

Symmetry code: (i)  $x, -y - \frac{1}{2}, z - \frac{1}{2}$ .

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and XP in SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DN2201).

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**supplementary materials**

*Acta Cryst.* (2007). E63, o3297 [ doi:10.1107/S1600536807030140 ]

## (*E,E*)-4,4'-Dichloro-2,2'-(1,1'-azinodiethylene)diphenol

J.-G. Chang, D.-F. Zhao, G.-F. He and J.-K. Li

### Comment

Recently, a number of azine compounds containing both a diimine linkage and N—N bonding have been investigated in terms of their crystallography and coordination chemistry (Kundu *et al.*, 2005; Kesslen *et al.*, 1999; Zheng *et al.*, 2005;). As an extension of work on the structural characterization of azine derivatives, the title compound, (I), was synthesized and its crystal structure is reported here.

In the title compound, there is a crystallographic centre of symmetry at the midpoint of the N—N bond (Fig. 1). The molecule displays the (*E, E*) conformation with respect to the C7=N1 and its symmetry related C7<sup>i</sup>=N1<sup>i</sup> double bond (Fig. 1). This configuration agrees with those commonly found in similar compounds (Glaser *et al.*, 1995; Hunig *et al.*, 2000). The crystal structure is stabilized by intramolecular O—H···N hydrogen bond and weak C—H···p interactions. (Table 1. and Fig. 2).

### Experimental

An ethanol solution (50 ml) of hydrazine (0.02 mol) and 1-(5-chloro-2-hydroxyphenyl)propan-1-one (0.04 mol) was refluxed and stirred for 5 h; the mixture was cooled and the resulting solid product, (I), was collected by filtration. Crystals suitable for single-crystal X-ray diffraction were grown by slow evaporation of a solution in 1,2-dichloroethane.

### Refinement

All H atoms were positioned geometrically and treated as riding on their parent atoms, with C—H(methyl) = 0.96 Å, C—H(methylene) = 0.97 Å, C—H(aromatic) = 0.93 Å, O—H = 0.82 Å and with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C}_{\text{methyl}}, \text{O})$  and  $1.2U_{\text{eq}}(\text{C}_{\text{aromatic}}, \text{C}_{\text{methylene}})$ .

### Figures

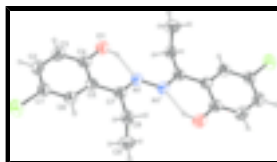


Fig. 1. The molecular structure of compound (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. Dashed lines show intramolecular hydrogen bonds. [Symmetry code: (i)  $1 - x, 1 - y, 1 - z$ ].

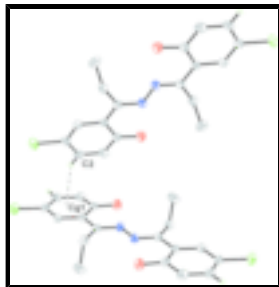


Fig. 2. View showing the C—H... $\pi$  interactio. H bond is shown as dashed line.

**(*E,E*)-4,4'-Dichloro-2,2'-(1,1'-azinodiethylene)diphenol**

*Crystal data*

$C_{18}H_{18}Cl_2N_2O_2$

$M_r = 365.24$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 9.3706$  (3) Å

$b = 13.9089$  (4) Å

$c = 6.7640$  (2) Å

$\beta = 103.021$  (1)°

$V = 858.92$  (4) Å<sup>3</sup>

$Z = 2$

$F_{000} = 380$

$D_x = 1.412$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 4397 reflections

$\theta = 2.2$ – $28.1$ °

$\mu = 0.39$  mm<sup>-1</sup>

$T = 273$  (2) K

Plate, yellow

$0.33 \times 0.23 \times 0.11$  mm

*Data collection*

Bruker APEXII CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 273$ (2) K

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan (SADABS; Sheldrick, 2003)

$T_{\min} = 0.882$ ,  $T_{\max} = 0.958$

9653 measured reflections

1507 independent reflections

1352 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.022$

$\theta_{\max} = 25.0$ °

$\theta_{\min} = 2.2$ °

$h = -10 \rightarrow 11$

$k = -16 \rightarrow 16$

$l = -8 \rightarrow 8$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.032$

$wR(F^2) = 0.089$

$S = 1.08$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0453P)^2 + 0.2176P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

1507 reflections  $\Delta\rho_{\max} = 0.17 \text{ e } \text{\AA}^{-3}$   
 111 parameters  $\Delta\rho_{\min} = -0.29 \text{ e } \text{\AA}^{-3}$   
 Primary atom site location: structure-invariant direct methods Extinction correction: none

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\text{sigma}(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	-0.00768 (5)	0.36471 (3)	1.03985 (8)	0.0643 (2)
O1	0.58588 (12)	0.37295 (8)	0.90005 (17)	0.0499 (3)
H1	0.5916	0.4059	0.8016	0.075*
N1	0.48939 (13)	0.47487 (9)	0.58463 (18)	0.0384 (3)
C1	0.16828 (19)	0.36915 (10)	0.9962 (2)	0.0433 (4)
C2	0.28142 (19)	0.32278 (11)	1.1277 (2)	0.0473 (4)
H2	0.2641	0.2898	1.2396	0.057*
C3	0.41983 (19)	0.32587 (11)	1.0916 (2)	0.0470 (4)
H3	0.4961	0.2947	1.1801	0.056*
C4	0.44800 (17)	0.37479 (10)	0.9253 (2)	0.0385 (3)
C5	0.33229 (15)	0.42224 (9)	0.7891 (2)	0.0351 (3)
C6	0.19278 (16)	0.41832 (10)	0.8310 (2)	0.0402 (3)
H6	0.1152	0.4496	0.7453	0.048*
C7	0.35581 (15)	0.47347 (9)	0.6084 (2)	0.0354 (3)
C8	0.23007 (16)	0.52225 (11)	0.4663 (2)	0.0419 (4)
H8A	0.1402	0.4879	0.4673	0.050*
H8B	0.2457	0.5198	0.3294	0.050*
C9	0.2138 (2)	0.62645 (13)	0.5253 (3)	0.0613 (5)
H9A	0.2073	0.6295	0.6648	0.092*
H9B	0.1265	0.6530	0.4409	0.092*
H9C	0.2973	0.6626	0.5076	0.092*

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0603 (3)	0.0673 (3)	0.0767 (4)	-0.0043 (2)	0.0394 (3)	0.0087 (2)
O1	0.0403 (6)	0.0577 (7)	0.0509 (7)	0.0051 (5)	0.0086 (5)	0.0132 (5)
N1	0.0386 (7)	0.0399 (6)	0.0387 (6)	0.0025 (5)	0.0127 (5)	0.0059 (5)

## supplementary materials

C1	0.0519 (9)	0.0366 (8)	0.0471 (9)	-0.0037 (6)	0.0229 (7)	-0.0017 (6)
C2	0.0658 (11)	0.0397 (8)	0.0399 (8)	-0.0032 (7)	0.0193 (7)	0.0032 (6)
C3	0.0567 (10)	0.0432 (9)	0.0388 (8)	0.0014 (7)	0.0059 (7)	0.0052 (6)
C4	0.0407 (8)	0.0346 (7)	0.0398 (8)	-0.0004 (6)	0.0083 (6)	-0.0022 (6)
C5	0.0394 (8)	0.0301 (7)	0.0369 (7)	-0.0012 (6)	0.0110 (6)	-0.0012 (5)
C6	0.0414 (8)	0.0371 (7)	0.0440 (8)	0.0018 (6)	0.0133 (6)	0.0019 (6)
C7	0.0368 (8)	0.0331 (7)	0.0367 (7)	-0.0003 (5)	0.0094 (6)	-0.0018 (5)
C8	0.0354 (8)	0.0501 (8)	0.0406 (8)	0.0004 (6)	0.0092 (6)	0.0068 (6)
C9	0.0636 (12)	0.0553 (11)	0.0637 (12)	0.0208 (8)	0.0116 (9)	0.0072 (8)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

C1—C1	1.7404 (17)	C4—C5	1.417 (2)
O1—C4	1.3410 (19)	C5—C6	1.400 (2)
O1—H1	0.8200	C5—C7	1.4738 (19)
N1—C7	1.2976 (18)	C6—H6	0.9300
N1—N1 <sup>i</sup>	1.393 (2)	C7—C8	1.504 (2)
C1—C6	1.373 (2)	C8—C9	1.520 (2)
C1—C2	1.381 (2)	C8—H8A	0.9700
C2—C3	1.374 (2)	C8—H8B	0.9700
C2—H2	0.9300	C9—H9A	0.9600
C3—C4	1.390 (2)	C9—H9B	0.9600
C3—H3	0.9300	C9—H9C	0.9600
C4—O1—H1	109.5	C1—C6—H6	119.4
C7—N1—N1 <sup>i</sup>	115.44 (14)	C5—C6—H6	119.4
C6—C1—C2	120.81 (15)	N1—C7—C5	116.02 (12)
C6—C1—C11	119.62 (13)	N1—C7—C8	123.55 (13)
C2—C1—C11	119.57 (12)	C5—C7—C8	120.41 (12)
C3—C2—C1	119.32 (14)	C7—C8—C9	111.89 (13)
C3—C2—H2	120.3	C7—C8—H8A	109.2
C1—C2—H2	120.3	C9—C8—H8A	109.2
C2—C3—C4	121.22 (15)	C7—C8—H8B	109.2
C2—C3—H3	119.4	C9—C8—H8B	109.2
C4—C3—H3	119.4	H8A—C8—H8B	107.9
O1—C4—C3	117.23 (13)	C8—C9—H9A	109.5
O1—C4—C5	122.97 (13)	C8—C9—H9B	109.5
C3—C4—C5	119.78 (14)	H9A—C9—H9B	109.5
C6—C5—C4	117.58 (13)	C8—C9—H9C	109.5
C6—C5—C7	120.52 (13)	H9A—C9—H9C	109.5
C4—C5—C7	121.90 (13)	H9B—C9—H9C	109.5
C1—C6—C5	121.28 (14)		

Symmetry codes: (i)  $-x+1, -y+1, -z+1$ .

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O1—H1 $\cdots$ N1	0.82	1.83	2.5513 (16)	146
C2—H2 $\cdots$ Cg1 <sup>ii</sup>	0.93	2.68	3.496 (2)	147

Symmetry codes: (ii)  $x, -y-1/2, z-1/2$ .

Fig. 1

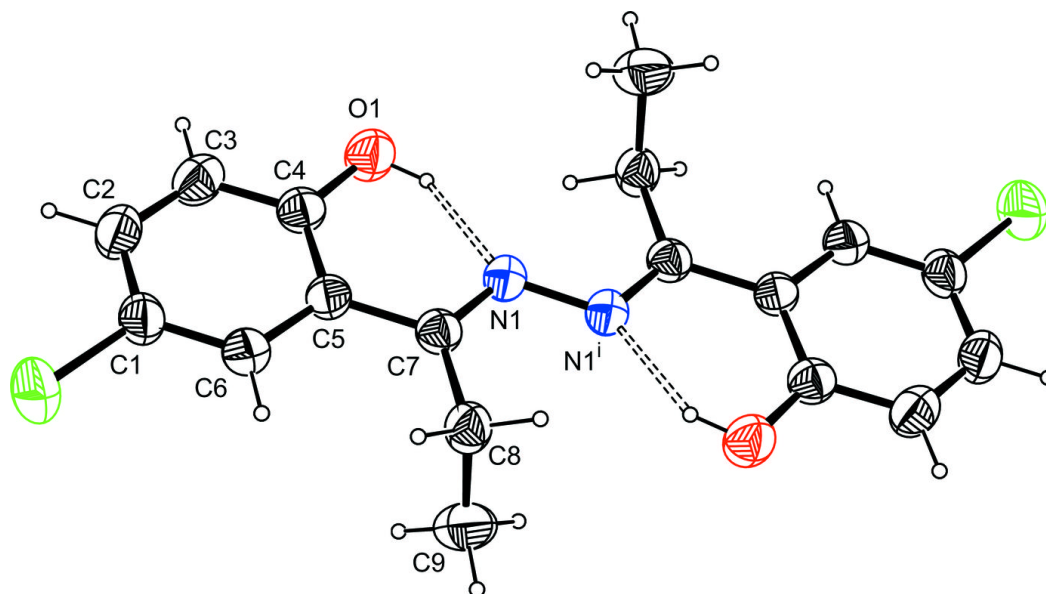


Fig. 2

